

Study of Different Parameters of Manmade Mohari Reservoir from Pathardi Tahsil, MS, India

Tilekar B.B.¹, Dhamak R.M.¹, Theurkar S.V.¹, Ghadage M.K.¹ and Patil S.B.²

¹JJT University, Jhunjhunu, Rajasthan-333001, INDIA

²Dept. of Zoology, Hutatma Rajguru Mahavidyalaya, Rajgurunagar, Tal. Khed, Dist. Pune, INDIA

Available online at: www.isca.in

Received 7th February 2013, revised 2nd March 2013, accepted 16th March 2013

Abstract

When water is polluted by various ways that means it affects flora and fauna of that area. The villagers mainly used it for irrigation and pisciculture activities. Water sample collected in Mohari dam located in southern part of Pathardi Tahasil, District-Ahmednagar (MS). Water is most precious natural resource expected to before from pollution. The physiochemical parameters are atmospheric temperature (AT), water temperature (WT), pH, electrical conductivity (EC), total dissolve solid (TDS), acidity (acid), alkalinity (alk), carbon dioxide (CO₂), dissolve oxygen (DO), were mentioned on monthly basis for period of one time annual cycle that is Jan 211 to Dec- 2011. The result revealed that the reservoir water is useful for human use.

Keywords: Irrigation, Mohari Dam, reservoir, parameters.

Introduction

Artificial dams are constructed in some rivers for providing irrigation to crop fields. Dam is constructed over a large area causes biodiversity loss in that area. Beside this inundation of adjacent lands withes over flowing water from dam sometimes causes great havoc to the inhabitatants and also causes loss of biodiversity in surrounding area of the dam. However there are no such studies from this region therefore present work is under taken. The present paper deals with the result of water analysis of manmade reservoir Mohari from Pathardi Tahasil of Ahmednagar District (MS), India.

The present study was conducted for one year that is Jan 211 to Dec- 2011 through the monthly sampling of Mohari reservoir. Mohari reservoir located in southern part of (19⁰ 9' N, 75⁰ 10' E) Pathardi Tahasil, which falls in Arangaon range of Balaghat, District-Ahmednagar. The reservoir is situated in southern part of Tahsil, which is hilly area with drought condition. The Mohari reservoir is Minor irrigation project type of reservoir near Mohari, about 9km from Pathardi Tahasil. It is constructed during the year 1973 having height of 14.35 meter. The catchment area is 9 square miles, which stores 72.50 mc.ft water and area under submergence is 75.30 hectares.

The density of diversity of zooplanktons is depending on water quality of reservoir. The zooplankton is microscopic free living floating organism, which occupy a central position between the autotrophy and other heterotrophs and from an important link in aquatic food web. Human life is living pattern without the presence of aquatic animals. All over the world, all fresh water habitats, lakes, ponds, reservoir, dams etc. The present paper deals with the result of water analysis of manmade reservoir

Mohari from Pathardi Tahsil of Ahmednagar District, Maharashtra State, India.



Figure-1
Location of Pathardi Tehsil, Ahmednagar (MS)



Figure-2
Location of Mohari reservoir, Pathardi

Material and Methods

The water samples were collected monthly. The physiochemical parameters are atmospheric temperature (AT), water temperature (WT), electrical conductivity (EC), total dissolved solid (TDS), acidity (acid), alkalinity (alk), carbon dioxide (CO₂), dissolved oxygen (DO), were carried out on field. Physical chemical characteristics of water were estimated following standard method¹.

Results and Discussion

The present investigatory study of Mohari Dam reveals all parameters are in favorable range for aquatic life, irrigation and domestic use. Present studies showed pH range favorable for aquatic life, irrigation and domestic use. The investigated results are as follows:

Temperature (AT and WT): Atmospheric temperature of surface water ranges from 24.6^oC to 43.5^oC during the study period. Minimum (24.6^oC) and maximum (43.5^oC) atmospheric temperature (AT) were recorded during winter and summer season respectively. The water temperature was maximum during summer (43.5^oC) and minimum during winter (24.6^oC). The result shows that water temperature varies with the atmospheric temperature, similar result were found Singhai S. et.al.², also found by a direct relationship between air and water temperature. During the summer season, solar radiations are and clear sky condition enhanced the atmospheric temperature. Where during the monsoon season, rainfall and cloudy-skies brought down the atmospheric temperature and subsequently the water temperature to minimum³.

pH: pH value of all samples lies in the range of 7.2 to 8.6 are slightly alkaline and suitable for irrigation purpose that is there is no alkalinity hazard (7.2- 8.1) during winter and higher value (8.1 to 8.6) during summer. Higher pH is normally associated with a high photosynthetic activity in water⁴⁻⁶. The pH of the water appears to be dependent upon the relative quantities of calcium carbonate and bicarbonates, being alkaline when disposal of wastes also bring about changes in the pH^{7,8}.

Electrical Conductivity (EC): Electrical conductivity (EC) is a measure of the salt content of water in the form of ion. EC value ranges from 189 µS/ cm to 294 µS/ cm with an average of 235µS/ cm. The month wise value shows the minimum during December. The concentration of EC increases during summer and reaches maximum in July. The increase in EC during pre monsoon period may be due to evaporation. This is in agreement with result obtained by Shankar P. et al.⁹.

Total Dissolve Solid (TDS): Total Dissolve Solid (TDS) are various kinds of mineral substances present in water. Some dissolved organic matter may also contribute to total dissolved solid. TDS value ranges from 105 mg/l to 169 mg/l. The season wise value shows the minimum during winter. The concentration of TDS in water gives an idea about suitability of

this water for various uses including potable water¹⁰. All the values of TDS were within the (500 mg/l) highest desirable limit¹¹.

Acidity (Acid): Acidity is found maximum during winter and minimum during summer. Acidity values of all samples lies in the ranges of 17.1 mg/l to 34.2 mg/l. Acidity of water is its quantitative capacity to react with a strong base to designated pH. Value of the acidity is about 200 mg/l¹² and observed values are far less than this, indicating that acidity of sample water is a safe range.

Alkalinity (Alk): Total Alkalinity shows seasonal variation in the study. Alkalinity value ranges from 215 mg/l to 284 mg/l. The values were high during the summer and low during winter. The fall in values during monsoon may be due to dilution of water. The high value of alkalinity indicates the presence of weak and strong base such as carbonate and hydroxide in the water body^{13,14}.

Carbon Dioxide (CO₂): The carbon dioxide level fluctuated between 1.8/l to 2.5mg/l. The seasonal value was 3.82 mg/l in winter, 2.27 mg/l in summer and 3.22 mg/l in rainy season. Low value of free CO₂ as observed during summer are mainly because of CO₂ is utilized in the polysynthetic activities¹⁴.

Dissolve Oxygen (DO): The dissolved oxygen is most important factor in fresh water life. In present study DO is ranged between 3.4 mg/l to 4.3 mg/l. The average DO value were 3.56 mg/l in summer, 4.05mg/l in winter and 3.95 mg/l during rainy season. The value of DO was obtained as following order, winter> rainy> summer season in present study. The results are similar and correlated with investigation of Dwivedi B.K.¹⁵. The phenomenon of re-oxygenation of water during monsoon may be due to the circulation and mixing by in flow water monsoon rains¹⁶. It further progressed in winter may be due to the circulation by cooling and draw down the DO in water.

Conclusion

The present study concluded that the Mohari manmade Reservoir indicates the higher values of some parameters of the samples. They minimize the suitability of these samples for drinking purposes without treatment. But, after the filtration and disinfection, naturally present impurities can be removed in water, which provide its suitability for drinking and domestic purposes. People depend on this water are often prone to health hazards due to polluted drinking water. Therefore, some effective measures are urgently required to enhance the drinking water quality by delineating an effective water quality management plan for the region.

Acknowledgements

Authors are thankful to the Shri. Dada Patil Rajale Shikshan Sanstha's, Hon. Rajivji Rajale, Rahulji Rajale and Secretary Shri. Pawar J.R. providing laboratory and library facilities to

complete this research work. I am also thankful to the Principal, Head, Department of Zoology, Hutatma Rajguru Mahavidyalaya, Rajgurunagar and the teaching and non teaching staff providing the necessary facilities to complete this research work. I wish to express my sincere thanks to Prof. Dr.

J.N.Nehul, Shri.Bambere P.V and Prof. Sanjay Bharate, Department of Botany for the thesis manuscript correction for the assistance in statistical analysis.

Table-1
Monthly reading in four sites of Mohari Reservoir (Jan to Dec. 2011)

Month	AT ⁰ C	WT ⁰ C	pH	Ec μs/cm	TDS mg/l	ACID mg/l	Alk mg/l	CO2 mg/l	DO mg/l
Jan	24.6	23.5	7.8	181	113	11.1	237	1.8	4.1
Feb	26.9	25.9	8.2	190	125	9.8	261	1.97	4.26
Mar	31.5	28.9	8.6	199	133	8.9	274	1.83	4.37
Apr	43.5	42.6	8.5	205	156	8.6	284	2.15	4.61
May	42.1	40.6	8.1	254	145	9.6	276	2.27	3.56
Jun	43.2	41.2	8.2	265	169	6.5	240	2.3	3.47
Jul	42.2	40.6	8.3	294	140	18.2	243	3.1	3.4
Aug	31.5	29.7	7.6	250	133	19.5	231	3.22	3.95
Sep	30.1	28.4	7.5	236	110	18.9	218	2.11	4.01
Oct	27.6	25.4	7.2	221	112	17.1	215	1.92	4.03
Nov	25.6	23.6	7.6	192	110	18.2	223	1.9	4
Dec	2.6.9	24.5	7.8	189	105	15.1	234	182	4.05

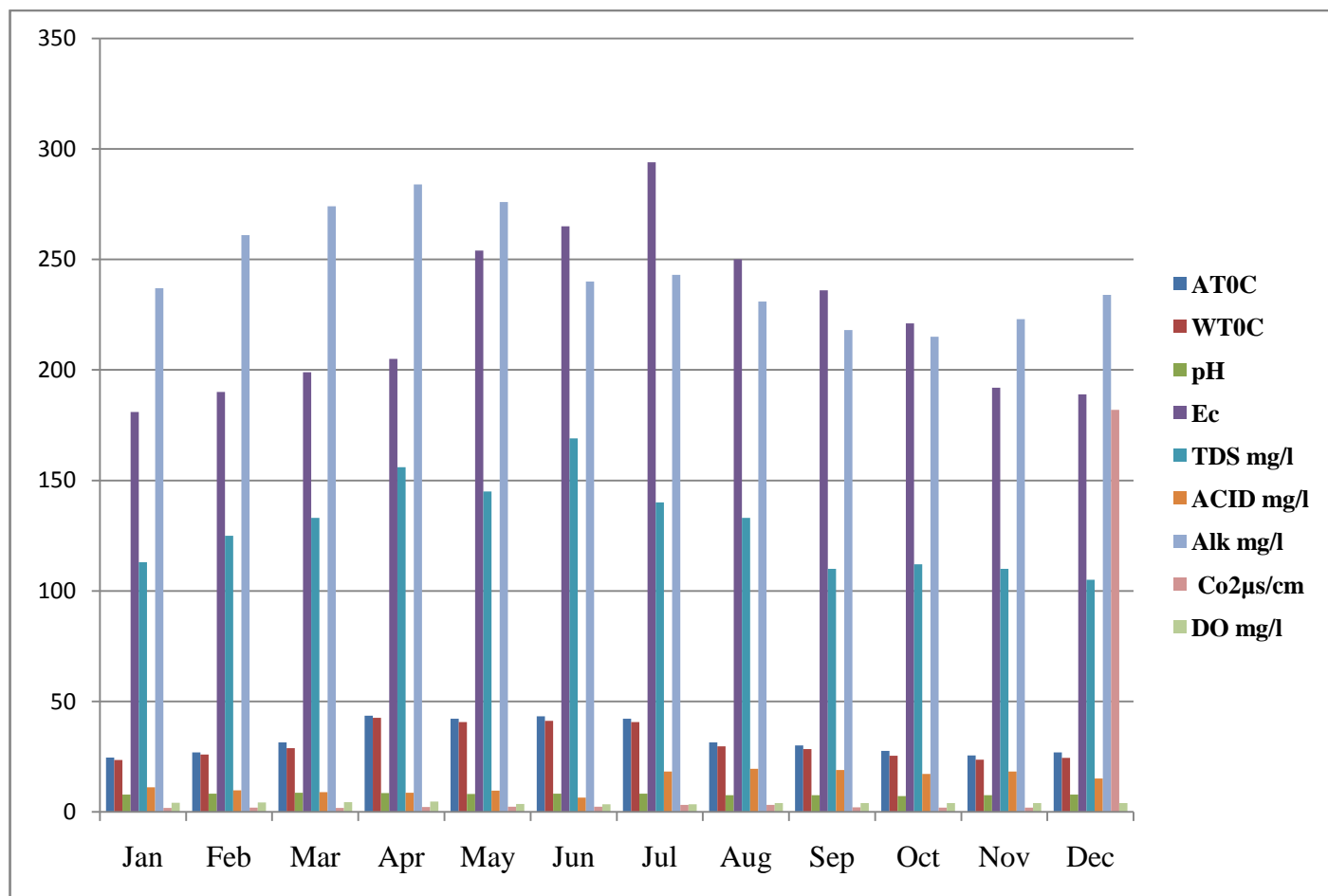


Figure-3
Monthly readings in four sites of Mohari Reservoir (January to December, 2011)

References

1. Trivedy R.K., Goel P.K. and Trisal C.L., Practical methods in ecology and environmental science Enviro media publication, Karad (India) (1987)
2. Singhai S., Ramani G.M. and Gupta U.S., Seasonal variation and relationship of different physiochemical characteristics in newly made Tawa Reservoir, Limnological (Berlin), *Poll Res J*, **21(1)**, 293-301 (1990)
3. Govindaswamy C. and Kannanm K., Rotifer of the pichavaram mangoves (Southeast coast of India) Hydrobiological approach, Mahasagar, *Bull. Natl. Inst. Ocenogr.*, **24**, 39-45 (1991)
4. King D.L., The role of carbon in eutrophication, *Ecology. Res J.*, **42**, 2035-2081 (1970)
5. Olsen R.D. and Sommerfeld M.R., The physiochemical limnology of desert reservoir, *Hydrobiologia*, **3J(2)**, 117-129 (1977)
6. Goel P.K., Trivedi R.K. and Bhave S.V., Studies on the Limnology of few fresh water bodies in southweterm Maharastra, India, *Res J. Environ, Pract*, **5(1)**, 19-25 (1985)
7. Pearsall W.H., Phytoplanktons in the English lake1, The production in the water of some dissolves substanscess of Biological importance, *Journal of Ecology*, **18**, 306-320 (1930)
8. Zafar A.R., Limnology of Hussian sagar Lake, Hydrabad, India phykas, *Poll Res J*, **5**, 115-126 (1966)
9. Shankar P., Jayaraman P.R. and Ganga Devi T., Studies on the Hydrography of the lotic ecosystem 'Killiar' Thiruvanthapuram, Kerala, India, *Poll. Res J.*, **21(2)**, 113-121 (2002)
10. Olsen R.D. and Sommerfeld M.R., The physiochemical limnology of desert reservoir, *Hydrobiologia*, **3J(2)**, 117-129 (1977)
11. WHO International standard for drinking water, Third edition, *WHO, Geneva* (1971)
12. Dwivedi P. and Sonar S., Evaluation of physiochemical and characteristics of water samples in water reservoir around Rono Hills, Doimukh (Dist.Papum pare), Arunachal Pradesh, *Poll. Res J.*, **23(1)**, 101-104 (2004)
13. Jain C.K., Bhatia K.S. and Vijay T., Ground water quality in coastal region of Andra Pradesh, *Indian Journal of Environ. Hith.*, **39(3)**, 182-192 (1997)
14. Sahai R. and Sinha A.B., Investigations on Bioecology of Inland of Gorakhpur (U.P.) India 1, Limnology of Ramgarh Lake, *Hydrobiologia*, **34**, 433-447 (1969)
15. Dwivedi B.K. and Pandey G.C., Physiochemical factors and algal diversity of two ponds (Girija kund and Maqubara pond) Faizabad, India, *Poll Res J.*, **21(3)**, 361- 369 (2002)
16. Hannam H., Chemical modification in reservior regulated Streams In: The ecology of regulated streams (Eds) Ward, J.W. and Stanford, J.A. *Edition plenum corporation publication*, 75-94 (1979)